Research Article



Synthesis and Biological Evaluation of 2-Substituted-Aryl- 3-Substituted- Phenyl- Oxazole- 4-Thiazolidines as Cyline-Dependent Protein Kinase 2 (CDK2) Inhibitors

VM Nikose^{1*}, MN Narule², KM Dadure³ and RU Thakre⁴

¹Department of Chemistry, Vidyabharti College, Seloo-442105, India

²Department of Chemistry, Vidyavikas Art, Commerce & Science College, Samudrapur, India

³Department of Chemistry, J. B. Science College, Wardha, India

⁴Department of Microbiology, Yashwantrao Chawhan Arts, Commerce & Science College, Lakhandur, India

***Corresponding authors:** VM Nikose, Department of Chemistry, Vidyabharti College, Seloo-442105, India, Tel: 9421804715, E-mail: research.vlsrc@gmail.com

Received Date: January 26, 2021 Accepted Date: February 26, 2022 Published Date: February 28, 2022

Citation: VM Nikose (2022) Synthesis and Biological Evaluation of 2-Substituted-Aryl- 3-Substituted- Phenyl- Oxazole- 4-Thiazolidines as Cyline-Dependent Protein Kinase 2 (CDK2) Inhibitors. J Org Chem Chem Sci 1: 1-16.

Abstract

A series of diverse heterocycles of 2-amino-4-phenyl thiazole (1a-o), 2-imine substituted phenyl-4-phenyl thiazole (2a-o) and 2- substituted- aryl- 3- substituted- phenyl- oxazole- 4-thiazolidines (3a-o) were prepared with excellent yield. The synthesized compounds were characterized by IR, H¹NMR, C¹³NMR & Mass spectral analysis. Induced-fit molecular docking (IFD) was performed on analogues of 2-Substituted-Aryl- 3-Substituted- Phenyl- Oxazole- 4- Thiazolidines against target protein (PDB ID 3EZR) having natural kinase inhibitor 3-methoxy-4-{3-[4-(4-methylpiperazin-1-yl)-1H-benzimidazol-2-yl]-1H-indazol-6-yl} aniline and synthesized derivatives were carried out against CDK2 using Auto dock tools. Further in vitro studies were evaluated for their antimicrobial and antifungal activities. These heterocycles have been tested against Gram-positive and Gram-negative bacteria and fungi and show outstanding therapeutic activity.

Keywords: 2- substituted - aryl- 3- substituted phenyl- oxazole- 4-thiazolidines; 2-imine; 4 phenyl thiazole; molecular docking; CDK inhibitors; Induced-fit molecular docking

^{©2022} The Authors. Published by the JScholar under the terms of the Creative Commons Attribution License http://creativecommons.org/licenses/ by/3.0/, which permits unrestricted use, provided the original author and source are credited.

Introduction

In the history of life, heterocyclic compounds have been utilised as colours, pharmaceuticals, and in many commercially important species and their analogues in which one or more ring carbons have been replaced by a heteroatom, such as nitrogen, oxygen, sulphur, phosphorus, silicon, or a metal, among other things. Nitrogen, oxygen, or both are present in the most prevalent heterocyclic systems. The most common and popular methods for preparing heterocyclic compounds is cyclization of suitable compounds. 4-Thiazolidinones are derivatives or Thiazolidine with a carbonyl group at the 4-position substitution is possible at 2, 3 and 5-position [1-2]. The nucleus is also known as the "wonder nucleus" since it produces a variety of derivatives with various biological activity. As a result, thiazolidinone with various substituents has been produced and used as a better medicinal agent in recent years. 4-oxo-thiazolidines are the most widely studied class of chemicals, with antibacterial, anti-inflammatory, anti-HIV, antitubercular, antioxidant, and analgesic properties [3-8]. The nucleus of 4-thiazolidinones has a distinctive role in medicinal actions such as antibacterial, anticancer, antiviral, cardiovascular, antitumor, and CNS depressive.

Experimental

The melting points were measured in uncorrected open capillary tubes. A Perkin-Elmer 157 spectrometer was used to measure IR spectra in KBr pellets. TMS was used as an internal standard in the recording of H1 NMR spectra in a CDCL3 on a Bruker- variah 300MHz FT NMR spectrometer. TLC on silica gel G plates was used to evaluate the purity of the compounds, and exposure to iodine vapours was used to find the spots.

Method of Synthesis

Synthesis of 2-amino, 4-substituted phenyl thiazole

In 100ml of ethanol, substituted acetophenone (0.1mole) and thiourea (0.2mole) were refluxed overnight with Br2 –H20. Other chemicals were also crystallised from DMF to produce a solid substance.

Synthesis of 2-imine substituted phenyl 4-substituted phenyl thiazole

A mixture of compound I (0.01mole) and substituted benzaldehyde (0.01mole) on discovery in ethanol in presence of Glacial acetic acid. The reaction mixture was refluxed for 1214hr and resulting solid was washed with ether and crystallized from DMF similarly other compound were also prepared.

Synthesis of 2-[4 oxo-2-substituted-aryl Thiazolidinyl] substituted phenyl thiazole:

In dry benzene, a mixture of compound 2 (0.01mole) and anhydrous ZnCl2 (one pinch) was added drop by drop with stirring, and the mixture was held at room temperature for 3 days before being refluxed for 12 hours. The resultant solid was washed and recrystallized from DMF after the react mixture was filtered and placed on to ice.

Result And Discussion

In light of these findings, various compounds containing 2-amino, 4-substituted phenyl thiazole, and 2-substituted phenyl imine were synthesized. Moiety has been connected to 4-substituted phenyl thiazole, 2-substituted aryl, 3-substituted phenyl thiazole, and 4-thiazolidinones.

Scheme-1 depicts the chemical sequence that leads to the synthesis of required heterocyclic compounds. By reacting substituted acetophenone with thiazole in the presence of Br_2 – H_2O and ethanol, the starting material 2-amino, 4-substituted phenyl thiazole (1a-o) was created. 2-Amino-substituted phenyl synthesis 2-[4-oxo-2-substituted aryl-Thiazolidinyl] substituted 4-substituted phenyl thiazole (2a-o) The IR, H1 NMR, C¹³NMR, and Mass spectra of the 2-substituted aryl-3substituted phenyl thiazole 4-thiazoldinones were obtained by reacting 2-amine substituted phenyl, 4-substituted phenyl thiazole with thioglycolic acid and zinc chloride in the presence of benzene.

Biological Studies

Norfloxacine and Griseofulvaline were used as standards in a biological study of thiourea with various acetophenones and (3a-o). The biological activity of compound (1) has been found to be higher than that of the freshly synthesised (3ao). For antibacterial and antifungal screening, the synthesised compounds were tested against staphylococcus aureus, *E. coli*, *P. vulgaris*, *A. niger*, *B. substillis*, and *Candida albicans* at a concentration of 100 mL as shown in Table 1.



0

5 T	R ₂
1)	H
2)	2-OH
3)	3-0H
4)	4-OH
5)	2-NO ₂
7)	4-NO 2
8)	2-Cl
9)	3CL
10)	3-OCH ₃
11)	4-OCH ₃
12)	3, 4, 5-(OCH ₃) ₃
13)	-N(CH 3)2
14)	2-Br
15)	4-Br



Compounds	S. aureus	B. substills	E. coli	C. albicans	A. niger
3a	+	+++	+	++	++
3b	++	++	+++	+++	++
3c	+	++	+++	++	++
3d	_	+++	+	+++	+++
3e	+++	++	++	+++	++
3f	++	+++	_	++	++
3g	+++	_	+	_	+++
3h	++	+	+++	++	_
3i	_	+++	+	+++	+++
3j	_	+++	+	+++	+++
3k	+++	++	++	+++	++
31	++	+++	_	++	++
3m	+++	_	+		+++
3n	++	+	+++	++	_
30	_	+++	+	+++	+++

Table 1: Antibacterial and Antifungal activities of compounds 3a-o

SM (streptomycin) and G F (Griseofulvin). The inhibition diameter in

Mm (-) < 6, (+) 7-9, (++) 10-15, (+++) 16-22, (++++) 23-28

Table 2. Characterization data of newly synthesized 2^{-1} + 0x0-2-substituted at yr minazondinyr substituted prenyr miazon $3/a^{-1}$

			Mol. Wt			0/	Analysis	found	
Comp	R*	Mol. Formula	(gm)	M.Pt (C ⁰)	RF Value	Yield	(Cal) %	(Cal) %	
							C I	I N	
3a	Н	$C_{18}H_{14}O_{2}N_{2}S$	322	165	0.91	83	67.08	4.34	8.69
							67.0	4.33	8.11
3b	2-OH	C ₁₇ H ₁₆ O ₂ N ₂ S	338	172	0.72	79	63.90	4.14	7.62
		17 10 2 2					63.2	4.74	7.43
3c	3-OH	C. H. O.N.S	338	181	0.75	75	63.90	4.14	7.62
		-17-16 - 2- 2					63.2	4.74	7.43
3d	4-OH	СНОМЯ	338	188	0.82	58	63.90	4.14	7.62
	_	17 16 2 2					63.2	4.74	7.43
							58.85	3.54	11.4
3e	2-NO2	$C_{18}H_{13}O_4N_3S$	367	191	0.77	54	58.84	3.41	11.0
							58.85	3.54	11.4
3f	3-NO2	C ₁ H ₁ O ₄ N ₂ S	367	191	0.77	54			
		10 13 4 5					58.84	3.41	11.0
							58.85	3.54	11.4
3g	4-NO2	$C_{18}H_{13}O_{4}N_{3}S$	367	191	0.77	54			
							58.84	3.41	11.0
							60.67	3.65	7.86
3h	2-CL	$C_{18}H_{13}O_{2}N_{2}SCI$	356	161	0.75	56			
							60.2	3.62	7.54

3i	4-CL	C ₁₈ H ₁₃ O ₂ N ₂ SCI	356	161	0.75	56	60.67 60.2	3.65 3.62	7.86 7.54
3j	3-OCH ₃	C ₁₉ H ₁₆ O ₃ N ₂ S	352	144	0.66	66	64.77 64.2	4.54 4.20	7.9 7.54
3k	4-OCH ₃	C ₁₉ H ₁₆ O ₃ N ₂ S	352	144	0.66	66	64.77 64.2	4.54 4.20	7.9 7.54
31	3-(OCH ₃) ₃	C ₂₁ H ₂₀ O ₃ N ₂ S	380	138	0.73	68	66.31 66.0	5.26 5.20	7.36 7.33
3m	-N(CH ₃) ₂	C ₂₀ H ₁₉ O ₂ N ₃ S	365	172	0.77	62	65.75 56.03	5.20 5.20	11.5 11.4
3n	2-Br	C ₁₈ H ₁₃ O ₂ N ₂ SBr	391	163	0.73	59	55.24 55.22	3.32 3.20	7.16 7.54
30	4-Br	C ₁₈ H ₁₃ O ₂ N ₂ SBr	391	163	0.73	59	55.24 55.22	3.32 3.20	7.16 7.54



Figure 1: Base molecule 2-Cyclohexa-1, 5-diphenyl-4-phenyl-[2, 3'] bithiazolyl-4'-one

compounds	IR(KBr)	H ¹ NMR (300MHz DMSO)	C ¹³ NMR (300MHz, DMSO-d6)
3a	2945 (C-H Aromatic stretch) 1791.8, 1713, 1641, 1520, 779 (C-S)	9.53 (S1H NH)	14.2,13.1,13.6,23.0,37.9,38.2, 34.5,39.4,40.0,58.5,76.8,7.3,111. 8,159.1,126.2,137.3,160.2,162.1.
3b	2945 (C-H Aromatic stretch),3275(OH of phenyl ring) 1639 & 1655 cy- clic carbonyl ring,690 (C-S-C link- age of thiazolidinone ring),1152 (c-o str)3209 (N-H Stretch) 1791.8,1713,1641,1520,779(C-S);	2.24,4.23,3.56 6.8-7.8 (M.8H Aromatic proton) 3.5 (s,2H, CH_2 Thiazolidine ring)	14.1,13.0,13.62,23.1,37.9,38. 0,34.1,39.2,40.1,58.3,72.3,7.3, 111.5,159.1,126.0,137.1,160.2,162.1.

3с	2945 (C-H Aromatic stretch) 3272(OH of phenyl ring) 1637 & 1653 cyclic carbonyl ring,693(C- S-C linkage of thiazolidinone ring),1151 (c-o str)3208 (N-H Stretch) 1791.8,1713,1641,1520,779 (C-S);	2.24,4.21,3.54 6.8-7.8(M.8H Aromatic proton)3.4(s,2H,CH ₂ Thiazolidine ring	14.2,13.1,13.2,23.0,37.8,38.1,34.2,39. 4,40.1,58.5,76.5,7.3,111.3,159.2,126. 1,137.32,160.2,162.2.
3d	2945 (C-H Aromatic stretch) 3274(OH of phenyl ring) 1638 & 1650 cyclic carbonyl ring,691(C- S-C linkage of thiazolidinone ring),1150 (c-o str) 3207(N-H Stretch) 1791.8,1713,1641,1520,779(C-S)	2.25,4.20,3.52 6.8-7.8(M.8H Aromatic proton)3.3(s,2H,CH ₂ Thiazolidine ring)	14.2,13.1,13.6,23.0,37.9,38.2,34.5,39. 4,40.0,58.5,76.8,7.3,111.8,159.1,126. 2,137.3,160.2,162.1.
3e	2945 (C-H Aromatic stretch)1791.8, 1713, 1641, 1520, 779(C-S)	2.22,4.20,3.51 6.8-7.8(M.8H Ar- omatic proton)3.3(s, 2H,CH ₂ Thiazolidine ring)	14.2, 13.1, 13.2, 23.1, 37.9, 38.1, 34.2,39.4, 40.0, 58.1, 76.8, 7.3, 111.2,159.0, 126.1, 137.3, 160.1, 162.1.
3f	2945 (C-H Aromatic stretch)1791.8, 1713, 1641, 1520, 779 (C-S)	2.20,4.21,3.54m, 6.8-7.8(M.8H Aromatic proton) 3.3(s, 2H, CH_2 Thiazolidine ring)	14.1, 13.2, 13.0, 23.0, 37.9, 38.1, 34.5,39.2, 40.0, 58.2, 76.8, 7.3, 111.2,159.1, 126.2, 137.3, 160.2, 162.1.
3g	2942 (C-H Aromatic stretch) 1791.8, 1713, 1641, 1520, 779 (C-S)	2.24,4.23,3.56 6.8-7.8 (M.8H Aromatic proton)3.4 (s,2H,CH ₂ Thiazolidine ring)	14.0,13.1,13.6,23.1,37.9,38.0,34.5,39.4, 40.0, 58.2,76.2,7.3,111.3,159.1,126.1,137.2,160.2,162.1.
3h	2944 (C-H Aromatic stretch) 1791.8,1713,1641,1520,779(C-S); 1689(C= O) of thiazolidinone ring	2.24, 4.23, 3.56 6.5-7.8 (M.8H Aromatic proton) 3.3 (s, 2H, CH_2 Thiazolidine ring)	14.0,13.213.6,23.0,37.9,38.2,34.2,39. 1,40.0,58.51,76.2,7.,111.3,159.1,126. 2,137.3,160.2,162.1.
3i	2941 (C-H Aromatic stretch) 1791.8, 1713, 1641, 1520, 779 (C-S);	2.24, 4.23, 3.56, 6.7-7.8 (M.8H Aromatic proton) 3.3 (s, 2H, CH ₂ Thiazolidine ring)	14.2,13.1, 13.5, 23.1, 37.9, 38.2, 34.5, 39.4, 40.1, 58.5,76.8,7.3,111.8,159.1, 126.2,137.3,160.2,162.1.
3j	2945 (C-H Aromatic stretch),3275(OH of phenyl ring) 1639 & 1655 cy- clic carbonyl ring,690 (C-S-C link- age of thiazolidinone ring),1152 (c-o str)3209 (N-H Stretch) 1791.8,1713,1641,1520,779(C-S);	2.24,4.23,3.56 6.8-7.8 (M.8H Aromatic proton) 3.5 (s,2H, CH_2 Thiazolidine ring)	14.1,13.0,13.62,23.1,37.9,38.0,34.1,3 9.2,40.1,58.3,72.3,7.3,111.5,159.1,12 6.0,137.1,160.2,162.1.
3k	2945 (C-H Aromatic stretch) 3272(OH of phenyl ring) 1637 & 1653 cyclic carbonyl ring,693(C- S-C linkage of thiazolidinone ring),1151 (c-o str)3208 (N-H Stretch) 1791.8,1713,1641,1520,779(C-S)	2.24,4.21,3.54 6.8-7.8(M.8H Aromatic proton)3.4(s,2H,CH ₂ Thiazolidine ring	14.2,13.1,13.2,23.0,37.8,38.1,34.2,39. 4,40.1,58.5,76.5,7.3,111.3,159.2,126. 1,137.32,160.2,162.2.
31	2945 (C-H Aromatic stretch) 3274(OH of phenyl ring) 1638 & 1650 cyclic carbonyl ring,691(C- S-C linkage of thiazolidinone ring),1150 (c-o str) 3207(N-H Stretch) 1791.8,1713,1641,1520,779(C-S)	2.25,4.20,3.52 6.8-7.8(M.8H Aromatic proton)3.3(s,2H,CH ₂ Thiazolidine ring)	14.2,13.1,13.6,23.0,37.9,38.2,34.5,39. 4,40.0,58.5,76.8,7.3,111.8,159.1,126. 2,137.3,160.2,162.1

3m	2945 (C-H Aromatic stretch)1791.8, 1713, 1641, 1520, 779(C-S)	2.22,4.20,3.51 6.8-7.8(M.8H Ar- omatic proton)3.3(s, 2H,CH ₂ Thiazolidine ring)	14.2, 13.1, 13.2, 23.1, 37.9, 38.1, 34.2,39.4, 40.0, 58.1, 76.8, 7.3, 111.2,159.0, 126.1, 137.3, 160.1, 162.1
3n	2945 (C-H Aromatic stretch)1791.8, 1713, 1641, 1520, 779 (C-S)	2.20,4.21,3.54m, 6.8-7.8(M.8H Aromatic proton) 3.3(s, 2H, CH ₂ Thiazolidine ring)	14.1, 13.2, 13.0, 23.0, 37.9, 38.1, 34.5,39.2, 40.0, 58.2, 76.8, 7.3, 111.2,159.1, 126.2, 137.3, 160.2, 162.1
30	2942 (C-H Aromatic stretch) 1791.8, 1713, 1641, 1520, 779 (C-S)	2.24,4.23,3.56 6.8-7.8 (M.8H Aromatic proton)3.4 (s,2H,CH ₂ Thiazolidine ring)	14.0,13.1,13.6,23.1,37.9,38.0,34.5,39.4, 40.0, 58.2,76.2,7.3,111.3,159.1,126.1,137.2,160.2,162.1



Figure 2: Natural Inhibitor 3-methoxy-4-{3-[4-(4-methylpiperazin-1-yl)-1H-benzimidazol-2-yl]-1H-indazol-6-yl} aniline of PDB ID 3EZR [9,10]

Function of CDK Activities

In the protein databank, there exist a lot of CDK2 electronic structures¹¹. Because of the existence of a natural inhibitor in one of the key active sties, 3EZR was chosen. This structure is also comprehensive and error-free. Inhibition of cyclin A and cyclin E-associated cyclin-dependent kinase-2 (CDK2) activity in tumour cells is an effective technique to induce apoptotic cell death via the E2F pathway. The cyclin groove recognition motif (CRM) in the CDK-inhibitory (CDKI) tumour suppressor protein p27 KIP1 was used to develop and synthesise a series of cyclic peptides with biological activity and structural characterization by NMR and X-ray crystallography¹². Inhibitors of cyclin-dependent kinases (Cdks) have been shown to block Cdk7 and Cdk9, which control transcription, in chronic lymphocytic leukaemia cells. Here we studied the novel Cdk inhibitor SNS-032, which exhibits potent and selective inhibitory activity

7

against Cdk2, Cdk7, and Cdk9¹³Overexpression of cyclooxygenase-2 (COX-2) is frequently observed in several human cancers, including lung, colon, and head and neck. Malignancies are also associated with the dysregulation of cell cycle events and concomitant elevated activity of cyclin-dependent kinases (CDK). CDK2 is a key cell cycle regulatory protein that controls the transition of cells from G1 to S phase. By suppressing CDK2 activity in the H358 human non-small cell lung cancer cell line, we provide many lines of evidence that demonstrate a functional role for CDK2 in IL-1-induced COX-2 expression [14].









Figure 3: Substituted 2-Cyclohexa-1,5-dienyl-4-phenyl-[2,3] bithiazolyl-4-one

FMolecular docking techniques are used to test fifteen molecules of 2-Cyclohexa-1, 5-diphenyl-4-phenyl-[2, 3'] bithiazolyl-4'-one for their ability to inhibit CDK2 (PDB ID 3EZR) enzyme. The goal is to look at the binding energies, different interaction postures, and probable hydrogen bonding of these compounds in order to better understand how efficient they are as CDK inhibitors, specifically CDK2 inhibitors.

Synthesized molecules (Ligand)

Table 4 listed fifteen compounds that were chosen to explore enzyme inhibition with manufactured molecules (also known as ligands).

Receptor enzyme

The electronic structure of CDK2, with PDB ID 3EZR, was chosen as a target protein. The protein file was obtained from an online database. 3-methoxy -4-{3-[4-(4-methylpiperazin-1-yl) -1H-benzimidazol-2-yl] -1H-indazol-6-yl} As a natural inhibitor, aniline¹⁷. The chosen enzyme structure was created without any uncertainties, such as missing atoms or amino acids. After removing all heteroatoms (non-receptor atoms like water, ions, and so on), Kollmann charges were assigned. Using AutoDock's Addsol function, the Solvation parameters were added to the final macromolecule structure¹⁸. The natural inhibitory site in an enzyme is considered as the active site of a chosen enzyme and used without further processing.

Г

	Surface Dot 3-D structure	Binding Energy	H-bond distance	Amino acids
		In Kcal/mole		
1*		-11.6863	2.744652	955-DEF 145-ASP 955- DEF 82-PHE 80-PHE 33-LYS
2		-10.7636		958- DEF
3		-10.0056	2.452949 2.900864	145 - ASP
4		-10.0197	2.633822 2.936493 2.482387 2.62280 2.478299	83-LEU

Table 4: Lists obtained binding energies for all docked molecules

5	And and a second s	-10.0197	2.895364	145- ASP
6		-10.3206	2.907496 2.552889	125 -HIS 145- ASP
7		-9.84859	2 . 5 5 5 3 0 1 , 2 . 8 9 2 5 2 7 , 2.900609A	14 -THR 127 -ASP
*8		-11.0653	2.980913A 2.22294A 2.735119A ,2.857214 A 2.5155	8 5 - G L - N,85-LEU

9			
10	-9.62396	,2.59475 ,2.593906 ,2.882631 .2.493378	83-LEU
11	-9.62396	,2.59475 ,2.593906 ,2.882631 .2.493378	83-LEU
12	-10.7636		

13	-9.68543		
14	-10.5341	2.840492 2.412427	
15	-10.475	1.064365 2.474559 2.296541 1.952081 1.935802 1.758240	ASP 145- PHE- 80 TYR- 15 TYR 15- LYS 33- ASP- 147

The binding energies for all docked compounds are listed in Table 4. The docking energies recorded range from -9.68543 to -11.0 653 kcal.mol1, which is higher than natural inhibitor data. The binding energy is not reported for a few compounds. All chemicals have a negative binding energy, which means they can form stable complexes. The development of hydrogen bonding improves the stability of a few molecules (HB). The findings of this study reveal that the current synthetic approach is a straightforward, efficient, and inexpensive way to make biologically active chemical (3a-o), with good results when evaluated at 100 mg Conc. against E. coli, S. aureus, P. vulgaris, A. niger, and C. albicans. Negative binding energies and compact inhibition are reported in molecular docking investigations of fifteen synthesised compounds, 2-Cyclohexa-1, 5-diphenyl-4-phenyl-[2, 3'] bithiazolyl-4'-one compounds. A number of them also mention hydrogen bonding as a possibility. And they're said to be the best inhibitors since they have greater ligand-enzyme interactions and are more stable. Therefore they show potency to be anti- CDKs agents. Their reported binding energies ranging from -9.68453

kcal. mol-1 to -11.6863 kcal.mol⁻¹ are reported in Table 4. The docking photographs in Table 4 show that the majority of the synthesised molecules bind in active site and remain within the boundaries of the designated active site.

Conclusion

Newly produced chemicals are analysed using elemental analysis, infrared spectra, C13 NRM, and H1 NMR spectra. There have been established efficient methods for synthesis of (3a-o) with high yield. It's also been reported that a molecule with a larger surface area that fits into the active site of a receptor enzyme has a lower binding energy value than a molecule with a smaller surface area.

Acknowledgement

The authors are thankful to Principal Dr. R.S. Bobhate, Vidya Vikas art, commerce and science college, Samudrapur for providing the laboratory facilities.

References

1. Perreux L, Loupy A (2001) A tentative rationalization of microwave effects in organic synthesis according to the reaction medium and mechanistic considerations. Tetrahedron 57: 9199-223.

2. Allen S, Newhouse B, Anderson AS (2004) Discovery and SAR of trisubstituted thiazolidinones as CCR4 antagonists. Bioorganic & Medicinal Chemistry Letters 14: 1619-24.

3. Athina A Geronikaki, Alexey A Lagunin, Dimitra I Hadjipavlou-Litina, Phaedra T Eleftheriou, Dmitrii A Filimonov, et al. (2008) J Medicinal Chemistry 51: 1601-9.

4. Walsh OM, Meegan MJ, Prendergast RM, Nakib TA (1996) Synthesis of 3-acetoxyazetidin-2-ones and 3-hydroxyazetidin-2-ones with antifugal and antifungal and antibacterial activity. Eur J Med Chem 31: 989–1000.

5. Abdel-Rahman RM (2001) Chemoselective heterocyclization and pharmacological activities of new heterocycles--a review. Part V-Synthesis of biocidal 4-thiazolidinones derivatives. Bollettino chimico farmaceutico 140: 401-10.

6. Shih MH, Ke FY (2004) Syntheses and evaluation of antioxidant activity of sydnonyl substituted thiazolidinone and thiazoline derivatives. Bioorg Med Chem 12: 4633-43.

7. Gadre JN, Nair S, Chitre S (2007) Synthesis of some new 4-thiazolidinones and thiazin-4-ones as biologically potent agent, Indian J. Chem., Sect. B: Org. Chem. Incl. Med. Chem 46B: 653-9

8. Vigorita MG, Ottanà R, Monforte F (2001) Synthesis and antiinflammatory, analgesic activity of 3,3'-(1,2-ethanedi-yl)-bis[2-aryl-4-thiazolidinone] chiral compounds. Part 10. Bio-org Med Chem Lett 11: 2791-94.

9. https://cdn.rcsb.org/etl/ligand/img/E/EZR/EZR-large. png

10. https://cdn.rcsb.org/etl/poseview/img/ez/3ezr/E/EZ-R/3ezr_EZR.png

11. Rose PW, Beran B, Bi C (2011) The RCSB Protein Data Bank: redesigned web site and web services. Nucleic Acids Res 39: D392-D401. 12. Andrews MJI, McInnes C, Kontopidis G, Innes L, Cowan A, et al. (2004) Design, synthesis, biological activity and structural analysis of cyclic peptide inhibitors targeting the substrate recruitment site of cyclin-dependent kinase complexes. Org Biomol Chem 2: 2735-41.

13. Chen R, Wierda WG, Chubb S (2009) Mechanism of action of SNS-032, a novel cyclin-dependent kinase inhibitor, in chronic lymphocytic leukemia. Blood 113: 4637-45.

14. Filardo EJ, Quinn JA, Bland KI, Frackelton AR Jr (2000) Estrogen-induced activation of Erk-1 and Erk-2 requires the G protein-coupled receptor homolog, GPR30, and occurs via trans-activation of the epidermal growth factor receptor through release of HB-EGF. Mol Endocrinol 14:1649-60.

Submit your manuscript to a JScholar journal and benefit from:

- Convenient online submission
- Rigorous peer review
- Immediate publication on acceptance
- Open access: articles freely available online
- High visibility within the field
- Better discount for your subsequent articles

Submit your manuscript at http://www.jscholaronline.org/submit-manuscript.php